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compounds 2. Valence
electrons 3. Noble gas
4. Cations 5. Anions 6.
Ionic 7. Crystalline 8.
negative/positive 9.
Positive/negative 10.
High 11. Large 12.
Stable 13. Molten 14.
valence electrons 15.
Metallic 16. Electrical
17. Malleable/ductile

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18. Ductile/malleable

19. body-centered/face-centered
20.

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Ionic Compound Formula Writing Worksheet Answers

Naming Ionic
Compounds Practice
Worksheet Name the
following ionic
compounds: 1) NH_4 ...

Review- Naming
Chemical Compounds
... Answers - Naming
Chemical Compounds

1) NaBr sodium
bromide 2) $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$ calcium acetate
3) P_2O_5

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diphosphorus

pentoxide 4) $\text{Ti}(\text{SO}_4)_4$

Naming Ionic Compounds Practice Worksheet

Chem 20 - Bonding

Review Worksheet

page 6 20. Explain,

using the metallic

bonding model, why

metals are malleable.

21. Define or describe

the following terms:

electronegativity

network solid lone pair

tetrahedral shape polar

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molecule ionic crystal
lattice polar covalent
bond orbital Answers –
next page

Bonding Review Worksheet - Ms. Mogck's Classroom

The Results for Type 1

Ionic Bonding

Worksheet Answers

Nomenclature

Worksheet. Practice

Worksheet. Ionic

Bonding Worksheet

Answers. Structure

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Type 1 Ionic Bonding Worksheet Answers Nomenclature ...

Ionic Bonding 1 Key
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work covalent bonds
and molecular.

Ionic Bonding Practice Worksheet Answer Key

CHAPTER 6 REVIEW

Chemical Bonding

SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1. a The notation for sodium chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule.

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(d) atom. 2. d In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b) positive ions. (d) negative ions.

6 Chemical Bonding

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Ionic Bonds Review

Sheet Answers Bonding

Review Station #2: 1.

Ionic bonds are formed

by a TRANSFER of

electrons. 2. Anions are

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attracted to a POSITIVE charge. 3. Melting

points of ionic compounds are typically (higher/lower) than those of covalent compounds. 4. What does it mean to say that an ionic compound is "brittle"?

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WORKSHEET: Ionic vs. Covalent! Ionic Bond between a Metal and

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Non-Metal (M + NM)

Covalent Bond

between a Non-Metal and Non-Metal (NM + NM) Determine if the elements in the following compounds are metals or non-metals. Describe the type of bonding that occurs in the compound.

WORKSHEET: Ionic vs. Covalent!

REMEMBE Ionic Bond Covalent ...

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Bonding Basics Review

ANSWER KEY 1.

Complete the chart using your knowledge of atoms. Element

Atomic Symbol Total # of Electrons # of

Valence Electrons # of Electrons Gained or

Lost Oxidation Number

Bromine Br 35 7 Gain 1

1-Lithium Li 3 1 Lose 1

1+ Calcium Ca 20 2

Lose 2 2+ Sulfur S 16 6

Gain 2 2-Boron B 5 3

Lose 3 3+

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Bonding Basics

Review Name -

Science Spot

NOMENCLATURE

REVIEW Molecular

Compounds, Ionic

Compounds, & Acids

ANSWER KEY 1. barium

sulfite 2. ammonium

phosphate 3.

phosphorus

pentabromide 4.

magnesium sulfate 5.

calcium oxide 6.

phosphoric acid 7.

sodium dichromate 8.

magnesium oxide 9.

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sulfur trioxide 10.

copper(II) nitrate 11.

hydroiodic acid 12.

dinitrogen monoxide

13. manganese(II)

oxide

NOMENCLATURE

REVIEW Molecular

Compounds, Ionic

Compounds ...

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Worksheet Answers

Best Chemical Bonds.

The fact that a molecule will react when it has an ionic bond is not new to us. It is something that all chemical scientists know already, and just as many professionals in other fields that deal with different ...

Worksheet Chemical Bonding Ionic and Covalent Answers Part 2

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