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Chemistry 11 Final Exam Study Guide Chemistry 11 - Final Exam Study Guide Page 18. An enthalpy diagram for an exothermic reaction: So, to summarize, in an exothermic reaction:  $\Delta H$  is negative Products are lower than Reactants on the Enthalpy Diagram Heat is released to the surroundings. In an endothermic reaction: Heat is absorbed from the surroundings and converted into enthalpy.

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(commonly called the General Chemistry Study Guide) This guide includes 201 pages of information and over 600 problems separated into first-term and second-term general chemistry material. Each section contains 8 chapters of material that also aligns to most general chemistry textbooks for a seamless addition to study materials for students.

### Student Study Materials | ACS Exams

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