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Chapter 4 Atomic Structure Guided Practice Problems Answers Chapter 4 Atomic Structure Section 4.2 The Structure of an Atom (pages 108–112) This section compares Page 6/23. Download Free Ch 4 Guided Atomic Structure Answersthe properties of three subatomic particles.

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Section 4.2 Structure of the Nuclear Atom One change to Dalton's atomic theory is that atoms are divisible into subatomic particles: Electrons

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Protons
Neutrons
11.

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CHAPTER 4 TEST: Atoms, Atomic Theory and Atomic Structure a student of Thompson, who tested the current theory of the atomic structure, using alpha particles (positively charged) which are directed at a sheet of gold foil, according to thompson's theory the particles should have passed through the gold with only slight deflection due to the positive charge in the foil, but he proved this wrong because majority of the particles passed through the gold...

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Section 4.2 Structure of the Nuclear Atom OBJECTIVES: Describe the structure of atoms, according to the Rutherford atomic model. 10. Section 4.2 Structure of the Nuclear Atom One change to Dalton's atomic theory is that atoms are divisible into subatomic particles: Electrons Protons Neutrons 11.

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Chapter 4 Atomic Structure Practice Chapter 4 Atoms Practice Problems 1) The primary substances of which all other things are composed are A) molecules. B) compounds. C) elements. D) electrons. E) protons. 2) Which of the following is a characteristic of the modern periodic table? A) A group is a horizontal row on the periodic table. B) A period is a

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